

Questions

- (8.1) A diesel fuel when tested has the same ignition characteristics as the mixture of 40% n- cetane and 60% hepta-methyl nonane. What is its cetane number? The fuel at 15° C has density of 825 kg/m^3 and mid-boiling point (T_{50}) of 240° C. Find the calculated cetane index of the fuel. How much error results in using CCI instead of CN?
- (8.2) What are the changes in volumetric efficiency for a gasoline (C_8H_{15}) engine when it is converted -by retro- fitment for operation on methane or hydrogen? Assume inlet conditions as 1 bar, 298 K and the engine size and geometry remain unchanged. Gasoline also enters the engine cylinder mostly as liquid.
- (8.3) Calculate energy content of 1 m^3 of stoichiometric mixtures with air of gasoline (C_8H_{15}), ethyl alcohol, methanol and hydrogen. Compare your results with those in Table 8. 13. Take standard conditions of 1 atmosphere (101 kPa) and 298K.
- (8.4) Rate the fuels methane, ethanol, gasoline, high aromatic gasoline, and diesel in terms of their potential to produce NO emissions based on adiabatic flame temperature data.
- (8.5) Calculate mass of CO_2 per MJ of energy for gasoline, diesel, propane and ethanol when burned as stoichiometric mixtures. Check your results with the data given in Table 8.13.
- (8.6) Find the contribution of 0.1% sulphur in fuel to PM as percentage of Euro 1 to Euro 4 heavy duty PM emission limits.
- (8.7) Discuss why Supreme Court of India could have ordered replacement of all the diesel buses by CNG buses in early 2003? How the CNG buses could meet those goals?