

## Introduction to Organometallic Chemistry

### A. G. Samuelson

#### **IV) Questions based on reactions with oxidation state change**

17. Oxidative addition & Vaska's complex
  18. Reductive elimination
  19. Reductive Elimination
  20. Oxidative coupling with C-C bond formation
  21. Metathesis reactions
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1. Complete the following equations giving the structure and electron count of the organometallic products
    - A.  $\text{MgBr}_2 + \text{K (metal)} \xrightarrow{\text{(THF solvent reflux)}}$
    - B.  $\text{MeCl} + \text{NaPb} \xrightarrow{\hspace{2cm}}$
  2. Suggest suitable methods for the preparation of
    - (a)  $\text{HCo(CO)}_4$
    - (b)  $\text{Na}_2[\text{Fe(CO)}_4]$
    - (c)  $[\text{CpRe(NO)(CH}_2\text{)(CO)}]^+$
  3.  $\text{CpCo(CO)}_2$  is a catalyst for synthesis of dimethylphthalate from acetylene and dimethylacetylene dicarboxylic acid. Why is dimethylphthalate the only product?
  4. Give an example of ring opening metathesis reaction using a suitable catalyst. How will you prevent polymerization in the a ring opening reaction using a metathesis catalyst?
  5. How will you maximise the formation of 1,5-hexadiene in the following reaction promoted by a metathesis catalyst ?
    - (a) cyclobutene and ethylene.  
What are the possible side products?